

**OXIDE CERAMIC NANOPARTICLES INTO FREE-FORMALDEHYDE WOOD ADHESIVE**

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**INTRODUCTION**

Despite the toxicity of formaldehyde, dangerous for the human health and environment, the use of resin based on formaldehyde as wood adhesive in the industrial production process of wood panels remains due to technical performance and economical reasons.

Synthetic resins free of formaldehyde have a higher price and are commonly used in specific types of products, such as pMDI.

Bio-based resins represents the only alternative for the production of free-formaldehyde resins non-derived from petroleum. These resins cannot provide enough hardness and moisture resistance in wood panels.

EcoPressWood European project is focused on the development of a bio-based free formaldehyde resin from residues of biodiesel production. To improve the hardness and moisture resistance of wood panels an extra reinforcement provided by oxide ceramic nanoparticles is analyzed.

Conventional oxide ceramic nanoparticles make a great commercial impact due to their low cost of production and that can be easily synthesized. Therefore they have been used to reinforce polymeric materials with improved properties.

Several studies have analyzed the use of oxide ceramic nanoparticles in the formulation of free-formaldehyde resins used as wood adhesives. The results of these nanofilled resins are promising and show a marked improvement over conventional resins.



**OBJECTIVE**

To analyze the effect of adding oxide ceramic nanoparticles into commercial free-formaldehyde wood adhesive resins for preparing particleboards. Analyze their stability into the resin and the effect in the final properties of the particleboards performed.

**EXPERIMENTAL PROCEDURE**

**MATERIALS** Acrodur 950 L, an acrylic formaldehyde-free binder for wood resin

Resin properties	
Chemical nature	Aqueous solution of a modified polycarboxylic acid
Solid content %	46
pH	3.16
Viscosity (mPa·s)	900-2500
Density (g/cm <sup>3</sup> )	1.2
Miscibility with water	Yes

Oxide ceramic nanopowder particles, Al<sub>2</sub>O<sub>3</sub>, ZnO, ZrO<sub>2</sub>, TiO<sub>2</sub>

Nanoparticle	Supplier	Particle size (nm)	Superficial area (m <sup>2</sup> /g)
Al <sub>2</sub> O <sub>3</sub>	lo-li-tec	20	<200
SiO <sub>2</sub>	Evonik	12	200
TiO <sub>2</sub>	lo-li-tec	10-25	50-150
ZnO	lo-li-tec	20	50
ZrO <sub>2</sub>	Tecnan	10-15	70-110



**EXPERIMENTAL TECHNIQUES**

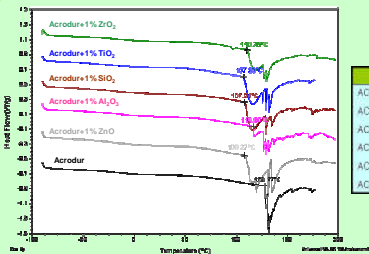
Nanoparticles at concentrations of 1-5%wt are added in the resin. Dispersions have been carried out by a high speed homogenizer (IKA RW20) at 2000 rpm during 2 h. The dispersions obtained have been followed in order to see their stability with time by their observance.

A Differential Scanning Calorimeter (Q200, TA Instruments) was used to determine the Tc (temperature to start the curing process) of the resin and the Tg (glass transition temperature). Conditions: drying of resin at 65°C for 60 min and ramp from -90 °C to 200 °C at 10 °C/min; inert atmosphere of N<sub>2</sub>(g).

Particleboards have been performed at lab scale (10x10 cm) with spruce particles (1.25-3.15mm), resin content of 23%/w/w, thickness 7.7 mm, 80 bar press, press time 78 s/mm and press T<sup>a</sup> 185 +/- 3°C. To characterize the wood panels samples tensile resistance or internal bond strength (EN 319), dimensional stability or thickness swelling (EN 317) and density (EN 232) have been measured.

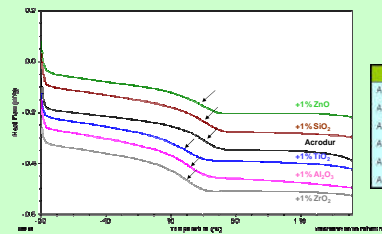
**RESULTS**

**Differential Scanning Calorimetry**



Sample	Tc (°C)
ACRODUR	129.2 ± 0.1
ACRODUR + 1wt% SiO <sub>2</sub>	107.9 ± 0.1
ACRODUR + 1wt% TiO <sub>2</sub>	107.2 ± 0.1
ACRODUR + 1wt% Al <sub>2</sub> O <sub>3</sub>	118.8 ± 0.1
ACRODUR + 1wt% ZnO	108.3 ± 0.1
ACRODUR + 1wt% ZrO <sub>2</sub>	110.8 ± 0.1

It is shown a decrease of the starting curing temperature with the addition of nanoparticles at 1wt%. TiO<sub>2</sub> produces the highest decrease (22°C less) while Al<sub>2</sub>O<sub>3</sub> the minor (11°C less) regarding the reference sample.



Sample	Tg (°C)
ACRODUR	361 ± 1
ACRODUR + 1wt% SiO <sub>2</sub>	32 ± 1
ACRODUR + 1wt% TiO <sub>2</sub>	20 ± 1
ACRODUR + 1wt% Al <sub>2</sub> O <sub>3</sub>	22 ± 1
ACRODUR + 1wt% ZnO	30 ± 1
ACRODUR + 1wt% ZrO <sub>2</sub>	20 ± 1

The addition of nanoparticles at 1wt% decreases the glass transition temperature. The least temperature obtained is produced with the addition of TiO<sub>2</sub> and ZrO<sub>2</sub>, and the highest with SiO<sub>2</sub>, regarding the reference sample.

**Dispersions**

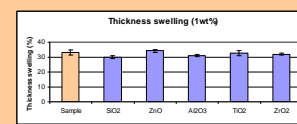
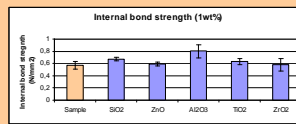


ZrO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> → settle down  
SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub> and ZnO → are stable

SiO<sub>2</sub> and ZnO → are stable around two weeks after stirring  
Al<sub>2</sub>O<sub>3</sub> and TiO<sub>2</sub> → settle down in few hours, dispersions are not stable

**Mechanical properties of particleboards**

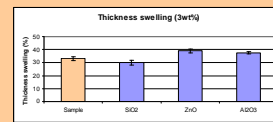
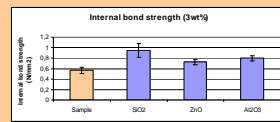
**1wt% of nanoparticles into the resin**



Nanoparticle	Density (kg/cm <sup>3</sup> )
Sample	670 ± 6
SiO <sub>2</sub>	660 ± 10
Al <sub>2</sub> O <sub>3</sub>	673 ± 3
TiO <sub>2</sub>	638 ± 23
ZnO	651 ± 8
ZrO <sub>2</sub>	671 ± 4

Improvement in internal bond strength (EN 319) and thickness swelling (EN 317) with the use of SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> at 1wt%

**3wt% of nanoparticles into the resin**



Nanoparticle	Density (kg/cm <sup>3</sup> )
Sample	670 ± 6
SiO <sub>2</sub>	657 ± 7
Al <sub>2</sub> O <sub>3</sub>	672 ± 3
ZnO	677 ± 10

Improvement in internal bond strength (EN 319) and thickness swelling (EN 317) with the use of SiO<sub>2</sub> at 3wt%

**CONCLUSIONS**

- Dispersions of SiO<sub>2</sub> and ZnO at 1, 3 and 5wt% in the acrylic resin at 2000 rpm for 2 hours showed to be stable with time
- The oxide ceramic nanoparticles selected decrease the temperature value required for starting of curing process and the glass transition temperature when they are added at 1wt% in the acrylic resin
- The study of the effect of nanoparticles on commercial acrylic resin show an improvement in the mechanical properties with the addition of SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> at 1wt%, and also SiO<sub>2</sub> at 3wt%.

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